An Unexpected Reaction Between 6,7-Dimethoxy-2*H*-1,3-benzothiazine and Dimethyl Acetylenedicarboxylate[#]

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Received August 24, 2005; accepted September 16, 2005 Published online December 13, 2005 © Springer-Verlag 2005

Summary. 6,7-Dimethoxy-2*H*-1,3-benzothiazine reacted with dimethyl acetylenedicarboxylate in aqueous methanol to give heterocycles with unexpected structures. A mechanism is proposed for the reaction.

Keywords. S,N-Heterocycles; Dimethyl acetylenedicarboxylate cycloadditions; 5,13-Dithiaxylopinine; Benzothiopyrane; Reaction mechanism.

Introduction

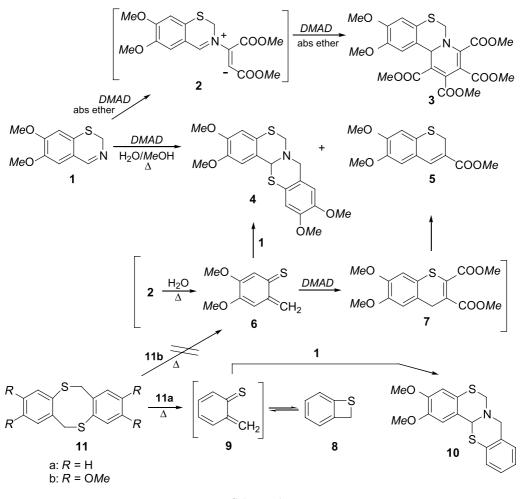
Earlier work has demonstrated that 6,7-dimethoxy-2*H*-1,3-benzothiazine (1) reacts with dimethyl acetylenedicarboxylate (*DMAD*) in anhydrous diethyl ether to yield the 1:2 adduct 3 by 1,4-dipolar cycloaddition [1]. However, it now turned out that the reaction of 1 with *DMAD* in refluxing aqueous methanol did not yield the expected product, but instead (\pm)-2,3,10,11-tetramethoxy-6*H*,8*H*,13*aH*-[1,3]benzothiazino[4,3-*b*][1,3]benzothiazine (5,13-dithiaxylopinine, 4) and the benzothiopyran derivative **5**.

Results and Discussions

The mechanism of this reaction has not yet been completely clarified. At present, we have only a tentative explanation for the formation of 4 and 5. In the presence

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[#] Dedicated to Professor Dr. András Lipták on the occasion of his 70th birthday



Scheme 1

of water, intermediate 2 may be transformed to the *ortho*-quinoidal form 6, which may react with 1 to furnish 4. Compound 5 may be formed from 6 by *DMAD* addition and subsequent decarboxylation and rearrangement of the adduct 7 (Scheme 1).

Several procedures are known for the generation of 9, *e.g.* from 6H,12H-dibenzo[*b,f*][1,5]dithiocine (**11a**) by flash pyrolysis, or from benzothiete (**8**) by heating above 90°C or by irradiation [2, 3]. Compound **8** can be used for various cycloaddition reactions. Facile thermal opening of the 4-membered ring confers on **8** the character of a heterodiene. We treated **8** with 6,7-dimethoxy-2*H*-1,3-benzothiazine (**1**) as dienophile and obtained (\pm)-2,3-dimethoxy-6*H*,8*H*,13a*H*[1,3]benzothiazino[4,3-*b*][1,3]benzothiazine (**10**) in a yield of 50%. The regioselectivity of the cycloaddition corresponds to those of the processes observed for other imines [4, 5].

The flash pyrolysis of 2,3,8,9-tetramethoxy-6H,12H-dibenzo[b,f][1,5]dithiocine (**11b**) has been investigated [6] under modified conditions. It emerged that the activation barrier for the reaction of the two *ortho* methoxy groups is lower than that for the cleavage of the 8-membered ring [7]. Further experiments are currently in progress.

Experimental

Melting points were determined on a *Kofler* apparatus and are uncorrected. Elemental analyses (C, H, N, S) were performed with a Perkin-Elmer 2400 CHNS elemental analyser, their results were found to be in good agreement ($\pm 0.2\%$) with the calculated values. Merck Kieselgel 60F₂₅₄ plates were used for TLC and Merck Silica gel 60 (0.040–0.063 mm) for column chromatography. 6,7-Dimethoxy-2*H*-1,3-benzothiazine (1) was prepared according to Ref. [8].

¹H and ¹³C NMR spectra were recorded in CDCl₃ solution in 5 mm tubes at rt, on a Bruker DRX 500 spectrometer at 500 (¹H) and 125 MHz (¹³C), using *TMS* as internal reference with the deuterium signal of the solvent as the lock. The MS spectra were run on a Finnigan TSQ 7000 mass spectrometer (Finnigan MAT Ltd, San Jose, USA).

(\pm) -2,3,10,11-Tetramethoxy-6H,8H,13aH-[1,3]benzothiazino[4,3-b][1,3]benzothiazine (5,13-dithiaxylopinine, **4**, C₁₉H₂₁NO₄S₂) and 6,7-Dimethoxy-3-methoxycarbonyl-2H-benzo[b]thiopyrane (**5**, C₁₃H₁₄O₄S)

Compound 1 (2.09 g, 10 mmol) was dissolved in aqueous methanol ($40 \text{ cm}^3 \text{ methanol} + 5 \text{ cm}^3 \text{ H}_2\text{O}$) and 1.42 g *DMAD* (10 mmol) were added. The solution was stirred under reflux. After 4 h, the reaction mixture was allowed to stand at rt overnight and the colourless radial needles that separated out were filtered off and washed with methanol (4, 0.96 g, 25%, mp 183–184°C, acetone). This product was identical in every respect with an authentic sample [9].

The mother liquor was evaporated to dryness, the residue was dissolved in the minimum amount of CHCl₃, and the solution was developed on a silica column. The elution was carried out with a 1% solution of methanol in CHCl₃. The eluted product **5** was recrystallized from methanol to yield 0.67 g (26%). Mp 115–116°C; ¹H NMR (500 MHz, CDCl₃): δ = 3.67 (d, 2H, 2-H), 3.78 (s, CO₂CH₃), 3.82 (s, OCH₃), 3.85 (s, OCH₃), 6.73 (s, 5-H), 6.74 (s, 8-H), 7.47 (s, 4-H) ppm; ¹³C NMR (125 MHz, CDCl₃): δ = 24.2 (C-2), 52.0 (OCH₃, ester), 56.0 (OCH₃), 56.0 (OCH₃), 109.8 (C-8), 113.3 (C-5), 120.1/124.2/126.3 (C-3, C-4a, C-8a), 137.2 (C-4), 147.2/150.6 (C-6, C-7), 166.5 (CO) ppm; EI MS (70 eV): m/z (%) =266 [M⁺] (87), 251 (100), 207 (53), 163 (21).

(\pm) -2,3-Dimethoxy-6H,8H,13aH[1,3]benzothiazino[4,3-b][1,3]benzothiazine

$(10, C_{17}H_{17}NO_2S_2)$

A mixture of 245 mg benzothiete (8) (2.0 mmol) and 628 mg 1 (3.0 mmol) was refluxed in 100 cm³ dry toluene for 5 h. The solvent was removed and the residue was purified by column chromatography (4×80 cm SiO₂, toluene:ethyl acetate 10:1). A pale-yellow product (335 mg, 50%) was obtained. Mp 157°C; ¹H NMR (500 MHz, CDCl₃): $\delta = 3.71/4.76$ (AB, ²*J* = 11.3 Hz, 2*H*, 6-H), 3.82 (s, OCH₃), 3.83 (s, OCH₃), 4.04/4.62 (AB, ²*J* = 17.0 Hz, 2*H*, 8-H), 6.09 (s, 13a-H), 6.59 (s, 4-H), 6.63 (s, 1-H), 6.59–7.12 (m, 4*H*, arom H) ppm; ¹³C NMR (125 MHz, CDCl₃): $\delta = 47.0$ (C-6), 55.8 (OCH₃), 55.9 (OCH₃), 57.8 (C-8), 65.9 (C-13a), 109.4 (C-4), 111.1 (C-1), 121.2/122.3/126.0/132.9 (C-4a, C-8a, C-12a, C-13b), 124.5/126.6/127.3/127.7 (C-9, C-10, C-11, C-12), 146.5/149.0 (C-2, C-3) ppm; EI MS (70 eV): *m/z* (%) =331 [M⁺] (30) [M–C₇H₆SH]⁺ (100).

Acknowledgements

We express our thanks to the Hungarian Research Foundation (OTKA) for grant OTKA T030647 and to the Hungarian Ministry of Health for grant ETT 55/2000.

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